

ILARA MOKIN, ONDO STATE



FACULTY OF BASIC AND APPLIED SCIENCES DEPARTMENT OF PHYSICAL AND CHEMICAL SCIENCES FIRST SEMESTER 2019/2020 B,Sc EXAMINATIONS

BCH 201- GENERAL BIOCHEMISTRY I

Time: 2 hrs Instruction: Attempt three (3) questions out of the five questions. Question 1 is compulsory and

1. (a) Define pH

carries 30 marks.

- (b) Why is pH important in Biology?
- (c) How can you determine the pH of a given solution?
- (d) Calculate the pH:
- (i) of 0.5M HCl solution?
- (ii)) If 100ml of 0.32M of NaOH are added to 200ml of 0.200M HCl, what is the pH of the resulting solution?
- (e) What are buffers?
- (f) Define Buffer Capacity
- (g) At what point does a buffer works best?
- (h) From the reaction, $HA \leftrightarrow H^+ + A^-$; Derive $pH = pKa + Log [A^-]/[HA]$.
- (i) A buffer solution contains 0.5M acetic acid and 0.5M sodium acetate. Calculate the pH of the buffer (pK₂ of acetic acid is 4.75).

30 marks

(2a) Classify the 20 amino acids based on their essentiality (2b) Describe the relationship between the molecular weight proteins	10 marks and number of residues 4 marks .6 marks	of
(3a) Giving correct example, write short notes on		

- (1) Epimers.....3 marks
- (ii) Enantiomers.....3 marks (iii) Anomers.....3 marks
- (iv) Diastereoisomers......3 marks

(3b) Draw the structures of Amylose and Amylopectin and state the differences between them8 marks	
(4a) Wi (i) (ii) (iii)	ith specific examples, different between Nucleosides and Nucleotides
(4b) Br (i) (ii) (iii)	iefly describe Palindrome
(5a) Gi	ve two (2) examples each of 3, 4, 5, 6 and 7 - carbon sugars5 marks
(5b) Wi (i) (ii) (iii)	ith the aid of annotated diagrams only, describe Enzyme
(5c) Wr (i) (ii) (iii) (iv)	rite short notes on the following; Active site